

WHAT IS MASS SPECTROMETRY



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Mass spectrometry (MS) is an analytical laboratory technique to separate the components of a sample by their mass and electrical charge. The instrument used in MS is called mass spectrometer. It produces a mass spectrum that plots the mass-to-charge (m/z) ratio of compounds in a mixture.

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Mass spectrometry (MS) is an analytical technique that ionizes chemical species and sorts the ions based on their mass-to-charge ratio. In simpler terms, a mass spectrum measures the masses within a sample. Mass spectrometry is used in many different fields and is applied to pure samples as well as complex mixtures.

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Mass Spectrometry. 1. The Mass Spectrometer In order to measure the characteristics of individual molecules, a mass spectrometer converts them to ions so that they can be moved about and manipulated by external electric and magnetic fields. The three essential functions of a mass spectrometer, and the associated components, are: 1.

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Mass Spectrometry Definition of Mass Spectrometry by

Recent Examples on the Web. The sophisticated tests now available include gas chromatography mass spectrometry, which can detect minuscule amounts of

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Note: All mass spectrometers that you will come across if you are doing a course for 16 - 18 year olds work with positive ions. Even if a few atoms in a sample of chlorine, for example, captured an electron instead of losing one, the negative ions formed wouldn't get all the way through the ordinary mass spectrometer.

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About Mass Spec ASMS

ASMS is pleased to provide information for those interested in learning more about Mass Spectrometry. We would like to expand the resources available on

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